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# Some aspects of thermal effects and application properties of engineering plastics and rubber  $\dot{r}$

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#### **Abstract**

The investigations presented here were made over the last few years at the department of Plastics and Rubber Engineering at the Polytechnic of Wuerzburg/Germany, exclusively in cooperation with German plastic and rubber processing or industrial applications companies. Practical aspects rather than basic research were stressed, e.g. faster characterization of material, characteristics of the final product, optimization of processing processes. Aspects of recycling, of growing importance today, were also considered.

*Keywords:* DSC; Plastic; Rubber; Thermosetting

## **1. High-performance plastics**

*1. 1. Non-reinforced and carbon-Jibre-(CF)-reinforced thermosets* 

1.1.1. Kinetic studies on epoxy (EP) resin systems based on differential *scanning calorimetry (DSC) measurements [I]* 

*Introduction.* Increasingly, aluminium alloys are substituted by fibre composite materials in the aircraft and aerospace industries. They excel because of their low weight at high strengths. However, thermoset matrix materials require long curing cycles during production and also extensive production planning. The matrices of the high-perform-

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a Types of reaction: 1st. 0, first-order reaction; nth 0, nth-order reaction; 1st 0 + autocat, first-order reaction with autocatalysis.

ante fibre composites used in practice are generally chemically crosslinked thermosets (EP resins, poly-imide (PI) resins, etc.), although high-temperature resistant (HT) thermoplastic materials are also being intensively investigated. Many tests with prototypes for optimizing the curing cycles still have to be made because the selection of process parameters, such as temperature, pressure, and viscosity, especially for complex parts or hybrid materials, leave only a small window open for the processor. To save development and production times and production costs, optimizing the curing process parameters by simulation is required. In addition to the reaction kinetic models based on DSC measurements, curing processes must be optimized by considering diffusion and heat transfer processes as well as the rheology in the simulation. But this will not to be discussed here. The systems investigated are listed in Table 1. The kinetic evaluations were made by the NETZSCH MultipleScan multistep kinetics  $[2,3]$  and according to ASTM E 698 [4]. In addition, degrees of conversion (curing, crosslinking) were determined by the residual heat method.

*Results.* Fig. 1 shows the typical shape of the DSC reaction curves for all systems investigated, except the fibredux system, and their shift towards higher temperatures with increasing heating rate (5, 10, and  $20^{\circ}$ C min<sup>-1</sup>) is shown for the Araldit system. It is evident that neither the reaction heat values,  $319 \pm 11.8 \text{ J g}^{-1}$  corresponding to a standard deviation of  $+3.7\%$  which is quite acceptable for such measurements, nor the peak heights,  $0.177 \pm 0.006 \text{ W g}^{-1}$  corresponding to  $\pm$  3.3%, show a significant dependence on the heating rate so that no competing or parallel reactions need be assumed for the kinetic model. During a second heating process, after a controlled cooling, of  $40^{\circ}$ C min<sup>-1</sup>, no exothermicity is observed. The weight losses after the second heating were less than 5%. The glass transition temperature at  $101.5 + 1.0^{\circ}$ C is also independent of the heating rate. This means that the curing of the samples, according to the DSC results, is completed after the first heating, regardless of the heating rate. Whether this is absolutely true would have to be checked by long-term curing because, according to the manufacturer, the glass transition of the fully crosslinked Araldit system should be at about 120°C. Therefore, except for the fibredux system whose DSC reaction curve is shown in Fig. 2 together with the second heating both kinetics used in this study should lead to similar results. But despite the unstructured DSC curves of the Araldit system, the best fit with the experimental data was reached by a three-step follow-on reaction model (Fig. 3). Therefore, the differences between the degrees of conversion calculated by both kinetic methods and by the residual reaction heat method are not surprising (Table 2). Similar results were found for all other systems (Table 3).

*Conclusions.* Comparing the results for the 5 investigated systems (Table 3), the following conclusions can be drawn.

1. None of the systems could be described by a single-step reaction of 1st order, e.g. according to ASTM E 698 when including the full DSC reaction peaks, but this was possible using MultipleScan kinetics even within the conversion range of 2-98% for each DSC curve. In addition, one should remember that ASTM E 698 uses only



Fig. 1. 1st heating DSC curves of the Araldit F system with heating rates of 5, 10 and 20°C min<sup>-1</sup>, from let to right



Fig. 2. First  $(-)$  and second  $(- - -)$  heating DSC curves of the Fibredux 914 system with a rate of  $10^{\circ}$ C min<sup>-1</sup>.



Fig. 3. Best MultiScan curve fits (solid curves) of the DSC reaction curves of Fig. 1 realized by a triple-step follow-on reaction.

discrete points of the reaction curve. This procedure was suggested by Carpenter [5] and applied in this investigation.

- 2. The degree of conversion evaluated by residual heat measurements as a dimension for the degree of cross-linking is only permissible for single-step reactions and/or if the reaction heats of the intermediate stages are negligible in contrast to the network-forming step, which can certainly not be assumed.
- 3. For comparing degrees of crosslinking, or better the relative degrees of network density, with reaction kinetic evaluations of DSC MultipleScan measurements, mechanical properties such as the interlaminar shear strength (ILSS) are the best, as shown in Fig. 4. But one should remember that ILSS [6,7], as compared to DSC, is a very time- and manpower-consuming method.



Summarized comparison with regard to the best MultiScan modelling results according to  $F$ -tests and correlation coefficient

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Fig. 4. Interlaminar shear strength and the degree of conversion of the Fibredux 914 C system calculated from MultiScan modelling of a triple-step follow-on reaction based on DSC curves with heating rates of 5, 10 and  $20^{\circ}$ C min<sup>-1</sup>.

*4.* Furthermore, the occurrence of negative shares of reaction steps in the MultiScan multistep model indicates that exothermal and endothermal reaction steps are superimposed within the entire reaction, that the measured reaction heat cannot correspond to the degree of conversion, and that the degree of conversion obtained by the multistep model must be higher than those obtained from the residual heat method or according to ASTM E 698 [8].

In conclusion, while considering only the best model calculated by the MultiScan multistep method within a conversion range of  $0.02-0.98$ , the differences relative to the residual heat method and the single-step lst-order reaction according to ASTM E 698 become evident because of the weak points of these two methods mentioned previously.

## 1 .1.2. *Influence of prepreg storage and laminate conditioning on the viscoelastic properties and the glass transition temperatures of EP- and PIICF-laminates*

The dynamic mechanical properties of these laminates are drastically influenced by prepreg storage and laminate conditioning, especially under humidity [9]. For this purpose, 8-layer EP/CF-laminates with a layer order of (O/45/-45/9O)s were made from UD prepregs of a thickness of about  $200 \mu m$  (trade name; Vicotex R 6376/40%/G 814 of Brochier S.A/Ciba-Geigy) with a resin content of 40%, and investigated by dynamic mechanical analysis (DMA) in the 3-point bending mode. This corresponds to an effective fibre content *F* of *60%* parallel to the long sample axis of the laminates.

The prepregs were stored for 0, 250 and 500 h at  $23^{\circ}$ C and 50% relative humidity. The laminates were produced in an autoclave according to the specification of the manufacturer  $\lceil 10 \rceil$ . Then, a portion of each laminate was conditioned for about 7 weeks at 70 $\rmdegree C$  in vacuum until a constant weight was reached, and another portion at 40 $\rmdegree C$  in a water bath. Fig. 5 shows the DMA curves of a laminate after vacuum drying without prepreg prestorage. The solid curve shows the storage modulus, the dash-dotted one the loss modulus and the dashed one the loss factor as a function of the temperature for a testing frequency of 1 Hz, a target deformation amplitude of  $\pm 30 \,\mu$ m and a heating rate of  $3^{\circ}$ C min<sup>-1</sup>, measured in a temperature range of  $-60$  to  $+320^{\circ}$ C. The effective sample dimensions were 40 mm in length,  $9.97 \pm 0.05$  mm in width, and  $1.78 \pm 0.18$  mm in thickness. The results are shown in Table 4. There is a striking decrease in the glass transition temperatures after water storage compared to the vacuum drying process, although this is a system optimized for a very low absorption of humidity  $[10]$ . Another interesting feature is the steps in the pattern of the storage modulus, at  $267^{\circ}$ C for the dry system and at 273.3"C for the system saturated with humidity. In addition, the wet

system shows a step in the tan  $\delta$  curve at about 195<sup>o</sup>C. Keeping in mind all these results as a function of prepreg storage at 23°C and 50% relative humidity, it can be seen that the characteristic temperatures and the appropriate storage moduli change significantly especially in the 2nd onset and/or in the tan  $\delta$  maxima. It must also be said that the storage moduli adjust to one another during the 500 h of storage. The same is true of the tan  $\delta$  values, although not to such a high degree. This indicates a certain dehydration of the parts saturated with humidity and a certain absorption of humidity during prepreg storage of the dried samples. The glass transition temperatures  $T_{\rm e}$ , however, are



Fig. 5. DMA spectrum (storage modulus  $(-)$ , loss modulus  $(- - -)$  and loss factor, tan  $\delta$  (-----)) of a (0/45/-45/9O)s EP/CF laminate made of Vicotex R 6376/40%/G814 UD prepregs after vacuum drying at 70°C without prepreg prestorage.



Influence of laminate conditioning and prepreg storage at  $23^{\circ}$ C and  $50\%$  relative humidity on the viscoelastic properties of EP/CF laminate

<sup>a</sup> Step due to softening of the sample during post-curing (DSC measurements).

<sup>b</sup> Due to storage conditions.

 $\epsilon$  tan  $\delta$  shoulder.

strongly influenced so that the permanent application temperature has probably to be restricted to 140°C under such extreme conditions. Up to this temperature, the material obviously tolerates considerable differences in humidity in its environment.

Similar results were obtained by TMA measurements (Dilatomer 402 ET, Netzsch-Gerätebau) of EP/CF and polyimide (PI)/CF laminates [11]. Both laminates were produced in an autoclave from  $32\frac{125}{\mu m}$  thick UD EP-prepreg layers (trade name: Fiberite E 1076 E of Dornier GmbH) with a resin content of 34–40%, as UD laminates and multidirectional (MD) laminates with a layer order of (45/90/-45/O/O/-45/90/45)2 s, and from 26 125  $\mu$ m thick UD PI-Roving prepreg layers (trade name: Sigrafil CI-1020 of Sigri Great Lakes Carbon GmbH; formerly Sigri GmbH) with a resin content of 33–37%, also as UD laminates and MD laminates with a layer order of  $\lceil 0(0/45/$ -45/0)3] s. The EP resin is probably a polyfunctional resin (maybe tetraglycidylmethylene-dianiline, TGDMA) with an aromatic diamine (maybe diaminediphenylsulphone, DDS) and the PI resin is probably a bismaleinimide with a diaminediphenylmethane hardener. Samples of size  $20 \times 8$  mm<sup>2</sup> were cut from laminate plates of dimensions  $300 \times 300$  mm<sup>2</sup> using a diamond-impregnated separating wheel under water cooling. The measurements took place immediately after storage with a heating rate of  $5^{\circ}$ C min<sup>-1</sup>. The length of the samples was 20 mm.

Table 5 shows the results for both matrices. Surprisingly, no reproducible results could be obtained for high effective fibre contents in the measuring directions. Microscopic inspections showed delaminations in the direction of the  $0^\circ$  fibre orientation caused by the sample preparation (drying and storage in humidity). This explains the non-reproducible results for the high effective fibre contents *F* of 86 or 100%. It is striking for both matrix resins how the glass transition temperature decreased con-

Table 4

Sample size in mm <sup>2</sup> /laminate type	Conditioning	<b>EP/CF</b> laminates			PI/CF laminates	
		F%	$T_{\rm e}$ /°C with 99% range of confidence	<b>Starting</b> temperature of post-curing with $99\%$ range of confidence	F%	$T_{\rm e}$ /°C with 99% range of confidence
$20 \times 8 /$ UD	Dry $70^{\circ}$ C/75% r.h.	100	n.r. n.r.	n.r. n.r.	100	n.r. n.r.
$20 \times 8$ /UD	$70^{\circ}$ C/95% r.h. Dry $70^{\circ}$ C/75% r.h.	$\Omega$	n.r. $208.5 + (-3.2)$ $125.5 + (-7.3)$	n.r. $237.9+/-2.0$ $186.0+/-7.5$	$\bf{0}$	n.r. $289.4+/-5.6$ $149.5+/-6.4$
$20 \times 8/MD$ parallel to	$70^{\circ}$ C/95% r.h. Dry $70^{\circ}$ C/75% r.h.	60	$103.0+/-2.1$ $216.7+/-4.1$ $143.1 + (-12.7)$	$177.7+/-3.5$ $227.7+/-3.0$ $175.4 + (-7.2)$	86	$125.2+/-7.9$ n.r. n.r.
$0^\circ$ fibre $20 \times 8/MD$ perpendicular to $0^\circ$ fibre	$70^{\circ}$ C/95% r.h. Dry $70^{\circ}$ C/75% r.h. $70^{\circ}$ C/95% r.h.	60	$126.1 + (-15.6)$ $217.9+/-2.9$ $146.3+/-11.9$ $123.2+/-5.1$	$157.9 + (-16.1)$ $230+/-6.3$ $174.7+/-7.0$ $158.2+/-5.1$	33	n.r. $303.3+/-14.3$ $154.5+/-10.8$ $141.5+/-7.0$

Table 5 TMA results

Key: r.h., relative humidity; n.r., non-reproducible results.

siderably on absorption of humidity. It is considerably higher than that of Vicotex R 6376/40%/G 814 EP/CF laminates, caused by the considerably higher humidity absorption of the matrix resins. This explains, of course, the extreme decrease of the glass transition temperature for a fibre content of 0% in the measuring direction.

# *1.2. Electron-beam-crosslinked polyamide 66 for applications in the electronic industry [12]*

For reasons of economic and application properties, the electronic industry, in contrast to the automotive industry where reactive compounding is of top interest, especially for thick-walled parts, prefers electron-beam crosslinking over chemical crosslinking. In many cases, this technology offers to replace very expensive and hard-to-process materials like PEEK by cheaper polymers like polyamide which are easy to process. Unfortunately, until now, there has been practically no information available concerning the electron-beam crosslinking of PA66. Therefore, Ultramid A3X165 nf of BASF AG, an important material for functional parts like contactors, was investigated here because if electric arcs develop at the moment of switching, the parts are subjected to temperatures up to 430°C. However, the melting point of noncrosslinked PA66 is only 255°C. So mobile parts of the contactors may melt so that contact carriers or slides will jam and the function of the contactor is no longer ensured. But it was thought that it would not be affected when those parts of the contactor are made of crosslinked material and that the required properties would only be achieved and maintained when a sufficient quantity of crosslinking material like triallylisocyanurate (TAIC) is added to the granulate. In this investgation, Betalink IC/W70 (trade name of Plastic-Technology-Service, Tauberzell, Germany), with a nominal TAIC content of 70 mass%, was used as a crosslinking agent. Using thermogravimetry (TGA), a value of  $67.8 + 0.2$  mass% was measured. In order to facilitate the handling, Betalink contains calcium silicate as a carrier. The next step was to find suitable inspection methods for the individual processing steps.

*Mixing step (adsorption of Betalink by the polymer).* First the polyamide was dried at a processing humidity of less than 0.2%. Then l-6 mass% of Betalink was added and the amount adsorbed by the polymer during mixing was determined. Mixing took place in a tumbling mixer at a speed of 60 r.p.m. for 15-20min. The warm granulate adsorbs the crosslinking agent up to 4 mass% so that it can practically no longer be seen after mixing. This procedure can readily be checked with a balance, especially if one works discontinuously in batch operation.

*Injection moulding step (weight loss during injection moulding).* Considering the fact that Betalink shows a weight loss starting at about  $100^{\circ}$ C, a loss of crosslinking agent must be expected during injection moulding. Injection moulding took place on an all-rounder 270-90-350 (Arburg, LoDburg/Germany). Because the viscosity of the batch is reduced by adding the crosslinking agent, cylinder and nozzle temperature had to be reduced by 10 $^{\circ}$ C from an addition of 5 mass% Betalink on. With 4 mass%, the temperature of the feeding zone was  $60^{\circ}$ C and both the cylinder temperature in all three zones and the nozzle temperature were 280°C. The mould temperature was 80°C. All the other injection moulding parameters were kept constant. TGA turned out to be the suitable method using the following program: sample mass, about 6.5 mg; purge gas and rate, nitrogen at 60 ml mm<sup>-1</sup>; heating up from 50°C at 10°C min<sup>-1</sup> to 80°C, holding for 5 min, and heating up again at  $10^{\circ}$ C min<sup>-1</sup> to 200°C, and holding again for 10 min.

Approximately 40min are required for one measurement. To reach the highest possible specific surface, the samples were carefully (avoiding heating) filed off from the injection-moulded parts. So the mass loss between 80 and 200°C including the isothermal holding turned out to be a suitable parameter for the characterization of the influence of the injection moulding step (Fig. 6, dash-dotted line). The blank value of the polyamide without the cross-linking agent was of course taken into consideration. It was independent of the crosslinking-agent content at the crosslinked material somewhat higher than at the non-crosslinked polyamide. This can be explained by polymer degradation by  $\beta$ -irradiation and, possibly, the slight residue of TAIC. The standard deviations of  $\pm 7.5\%$  for 1% Betalink and  $\pm 2.6\%$  for 6% Betalink determined in 9 single measurements do not indicate accidental inhomogeneity of the samples regarding a sample mass of about 6.5 mg. On the basis of the TGA results, the weight loss of 2g samples held for  $60 \text{ min}$  at  $200^{\circ}$ C in an oven was determined. Fig. 6 (solid line) shows a good correlation between TGA and oven values.



Fig. 6. Triallylisocyanurate (TAIC) content after injection moulding (TGA and oven storage) and after electron-beam crosslinking (TMA and soldering iron test).

This means that only an oven is needed at the injection moulders to monitor this production step.

*Influence of crosslinking.* The crosslinking took place in a 4.5 MeV electron beam accelerator. 1000 kGY (1 Gy = 1 J kg<sup>-1</sup>) is regarded as the optimal beam dosage for polyamide without crosslinking agent [13]. With Betalink, we found by extraction measurements that 100 kGy are enough. Since mechanical methods such as tensile strength are very time-consuming, especially when the temperature dependence is of interest, as is the case here, dilatometric penetration measurements were carried out based on the determination of the Vicat softening temperature. The hemispherical pin (radius, 0.5 mm; height, 2 mm) of a penetration sample probe was loaded with 100 g and then the following program was run:

*Heating at 10°Cmin<sup>-1</sup> from 200 to 270°C followed by 10 min isothermal holding.* The difference in the penetration depths in the isothermal part of the penetration curve at *270°C* between 2 and 10 min was defined as the suitable parameter. A clear correlation between the adsorbed Betalink content and the penetration value which characterizes the crosslinking density is apparent up to 4% Betalink (Fig. 6, dashed line). On the basis of these results, a soldering iron penetration tester was developed which can be used simply on the production line. These results are also entered in Fig. 6 (dotted line) with a temperature of about  $340^{\circ}$ C for the penetration pin, a load of 22 N and a penetration time of 10 s. Furthermore, it could be shown by gel permeation chromatography that the medium mol mass of the soluble share of the crosslinked samples decreased when the crosslinking agent increased, which means that there is a correlation between mol mass decrease and the inverse penetration depth.

#### **2. Blends and copolymers**

Impact-modified polypropylene blends and polypropylene/polyethylene block copolymers are becoming very important materials for automobile applications, especially because of their favourable cost-performance ratio and their good recycling properties [14].

Therefore the target of this study was to establish a correlation between the time-consuming impact strength measurements by instrumental puncture tests, according to DIN 53 443, and values determined considerably more quickly by DMA within the glass transition range of the impact-modifying EPM component around  $-60^{\circ}$ C. Table 6 lists the investigated materials, which differ with respect to their type (blend or block copolymer) as well as their softphase content and the surface structure of the injection-moulded parts. The blends are so-called reactor blends.

In addition to the results of the instrumental puncture and the DMA tests, DSC and melt-flow-index (MFI) measurements were carried out. The results are summarized in Table 7 and compared with the data given by the raw material manufacturers. The reasons for the deviations between the manufacturer's data and the values calculated from the DSC measurements, as observed in part for the soft phase, is probably due to the separation of the superimposing PE and PP melt peaks (only partial area integration but no peak separation software was available), difficulties in the definition



Table 6 Investigated plastics

a Incomplete supplier data (experimental stage product).

<sup>b</sup> Only one surface of the sample.

No.	MFI results/ $(g/10 \text{ min})$		Weight% calculated from DSC measurements			Weight% softphase <sup>d</sup>
	Measured values	Values given by the supplier <sup>c</sup>	<b>PE</b>	PP	<b>EPM</b>	
1	$4.3 \pm 0.12$	4.5	1.7	66.7	11.6 <sup>d</sup>	$8 - 10$
$\overline{c}$	$2.5 + 0.09$	1.5	3.8	84.6	11.6	$12 - 15$
3	$5.3 \pm 0.18$	4.0	4.2	83.9	11.9	$13 - 17$
4	$12.6 \pm 0.69$	16.0	$\bf{0}$	74.2	5.8 <sup>d</sup>	$17 - 20$
5	$3.6 \pm 0.15$	$5 - 8.5$ <sup>a</sup>	16.9	43.1	40.0	$28 - 30$
6	$4.8 \pm 0.15$	$3 - 4^a$	11.5	52.6	36.9	$40 - 45$
7	$7.7 + 0.33$	$-b$	14.0	33.7	52.3	$45 - 50$
8	$3.3 \pm 0.14$	$1.4 - 2b$	15.6	60.7	23.7	$45 - 50$
9	$1.4 \pm 0.05$	2.0	16.2	40.4	43.4	$45 - 50$

Table 7 MFI and composition of the blends/copolymers

a MFI 190/21.6.

<sup>b</sup> No supplier values available (experimental stage product).

' Hostalen PP leaflet.

d 20% talcum reinforced.

of the base line especially at very low PE contents, and the empirical equation for the evaluation of the PE content, which is only valid up to a PE content of approx. 20%  $[15]$ .

Because the soft phase content stated by the manufacturer is made up of EPM and PE, these two values have to be added to Table 7 in order to be comparable with the manufacturer's data. One of the most interesting properties regarding exterior plastics parts in car manufacture is the low temperature flexibility at impact stress. The energy absorbed by the sample at the glass transition temperature and a puncture speed of  $4.43 \text{ m s}^{-1}$  is considered to be a standard test method for estimating the tough/brittle behaviour of plastics exposed to impact stress. Starting at room temperature, the temperature was raised or lowered in steps of 5°C until the transition temperature of the tough/brittle zone was determined. Three samples per material were punctured at each temperature. As this test procedure requires a lot of time and equipment, other methods were looked for. DMA seemed to be a suitable alternative. Fig. 7 shows a typical three-point bending DMA spectrum of the investigated blends with the shift of the glass transition temperatures characterized by the onset temperatures of the storage modulus and the peak temperatures of the loss factor tan  $\delta$  for five frequencies between 1 and 20 Hz and a heating rate of  $3^{\circ}$ C min<sup>-1</sup>. The values for the three phases of the blend at 1 Hz were calculated with  $-140.8^{\circ}$ C for the onset and  $-131.7^{\circ}$ C for the tan  $\delta$ peak of the polyethylene component, with  $-65.6$  and  $-55.7$ °C for the EPM component, and  $-3.7$  and  $13.0^{\circ}$ C of the polypropylene component. Using the Arrhenius equation, which was not done here, one can also calculate the glass transition temperatures at application frequencies that cannot be reached by the DMA spec-





trometer. This is also very important for application of finished plastic parts, for example in the automotive industry. Since it can be assumed that the EPM component is mainly responsible for the low-temperature flexibility, the peak height  $\Delta \tan \delta$  and the peak area were evaluated. Taking into consideration that the transition temperatures from the puncture tests were determined in steps of  $5^{\circ}$ C, satisfactory correlations between the soft phase contents determined by DSC as well the peak heights tan  $\delta\Delta$  of the EPM peaks and the transition temperatures could be established for the unfilled as well as for the talcum-filled blends and block copolymers, with the exception of one polymer blend (material no. 6) (Fig. 8). The expected strong influence of talcum is striking. Surprisingly, greater uncertainty is to be observed with the area evaluation than the peak height evaluation. To sum up, it can undoubtedly be said that DSC and also DMA appear to be suitable for distinguishing elastomer-modified PP blends with respect to their low-temperature flexibility. These investigations are being continued, particularly in view of the influence of the surface structure on the transition zone determined with the puncture test, especially since no strong deviations occur when grained materials are left out.

In this context, the presentation of studies on the anisotropy of self-reinforcing liquid crystalline polymer blends for injection moulding [16] or on the optimization of the heat stability of thermoplastic elastomers based on urethane by means of blending with thermoplastics such as polybutylene terephthalate, polyphenylene sulphide or aromatic polyester carbonate, or of thermoplastic elastomers which are based on ether or ether ester amide [17], is also of interest from the scientific point of view. These materials are not yet in general use, so they will not be discussed in this article.



Fig. 8. Tough/brittle transition temperatures evaluated by the instrumented puncture test in comparison to the  $\Delta$  tan  $\delta$  peak values and to the content of the EPM component.

#### **3. Rubber composites**

Today, rubber composites are considered high-performance materials that are used for a variety of industrial applications. A comfortable climate in the interior of a car, irrespective of the time of year or kind of weather, is only possible with air conditioning. In Europe, the number of cars equipped with air conditioning has steadily risen in recent years. Depending on the car model, between approx. 30%, e.g. Mercedes-Benz middle class, and approx. 90%, e.g. Mercedes-Benz S-class, of cars are provided with air conditioning [ 181. Flexible hose connections are indispensable for the compensation of the relative movements of the units positioned between motor and body and for the necessary workability during installation. Apart from many other requirements, e.g. made with respect to the low temperature and heat pressure resistance, thermal expansion behaviour, the tensile and flexural rigidity or the bursting pressure, vibration (noise production) and impulse pressure behaviour, to name just a few [19], these hoses must be resistant and impermeable to the coolant employed so as to ensure reliable operation of the unit and protection of the environment. Due to its ozone-depleting potential (ODP), difluorodichloromethane(R 12) used in the past, had to be substituted by the chlorine-free trifluorofluoroethane (R 134a). Since this coolant possesses a global warming potential (GWP), however, new coolant hoses which prevent coolant diffusion into the environment had to be developed. This problem can only be solved by a multilayer hose construction.

Because of this new coolant, a new material evaluation had to be made to ensure that, among other demands [19], no moisture could penetrate into the coolant circulation, which would not only lead to corrosion damage of the air-conditioning device but also to frosting-up in the thermostatic expansion valve in particular. The water (vapour) permeability was investigated by a device developed by the German Institute for Rubber Technology in Hannover. The quantitative determination of the penetrated water was effected by gas chromatography through measurement of the acetylene, which is produced quantitatively and stoichiometrically from water and calcium carbide. Four commercially available hoses were tested. Their details are described in Table 8. Fig. 9 displays the results. Hoses 1 and 4 were the best by far, which was to be expected on account of the butyl rubber layer. The bad result of Hose 3 came as

	Hose no. 1	Hose no. $2$	Hose no. $3$	Hose no. 4			
Thermoplastic barrier film $PA_6 + 12$							
	impact modified	PA 6	PA 6	PA $6 + 12$			
Elastomeric inner layer	<b>HR</b>	<b>NBR</b>	<b>HNBR</b>	<b>IIR</b>			
Elastomeric middle laver	<b>HR</b>	none	none	none			
Fabric	Polyester	Rayon	Aramid	Polyester			
Elastomeric outer layer	<b>EPDM</b>	<b>CR</b>	<b>HNBR</b>	<b>EPDM</b>			

Table 8 Investigated hose designs



Fig. 9. Water (vapour) permeation rates through coolant hoses of air-conditioning devices at 70°C versus time.

**a** surprise, since it was assumed that the relatively expensive hydrogenated nitrile rubber (HNBR) would be a very good barrier against moisture.

To gain further insights into the changes in mechanical properties, especially those dynamic mechanical properties which determine the vibration and noise production behaviour, DMA spectra were run with the best (hose no. 4) and the worst (hose no. 2) hose after they had been subjected to the water (vapour) permeability testing device to 100% moisture in a waterbath of 70°C for 9 days (216 h).

Decisive for a good noise reduction behaviour is that almost no changes occur in the operating temperature range between  $-30$  and 150°C. Looking at the characteristic peaks in the tan  $\delta$  curve (Table 9) for the best (no. 4) and the worst (no. 2) hose, it becomes clear that hose no. 4 meets this prerequisite. An important fact also seems to be that there are no changes in the tan  $\delta$  values of hose no. 4 within the entire operating temperature range from  $-30$  to  $+150^{\circ}$ C, whereas this is only true above 0°C for hose no. 2. At this temperature, however, hose no. 2 has a tan  $\delta$  value of 0.2 which is better than that of hose no. 4 (tan  $\delta = 0.1$  at 0°C). As the temperature increases, their values assimilate. This leads to the conclusion that hose no. 4, after being in service for a prolonged period of time, shows less noise production in winter in the cold start phase of a car than hose no. 2.

Using IR spectroscopy, no significant changes could be found.

#### 4. **Recycling**

Finally, the subject of recycling has to be addressed. When discussing the recycling of polymer materials, one has to distinguish between three large groups: the material, the raw material or chemical, and the energetic or thermal recycling. With a few exceptions

	Hose no. 2	Thickness/mm	Hose no. 4	Thickness/mm
Thermoplastic barrier				
film	<b>PA 6</b>	0.1	$PA6 + 66$	0.4
Elastomeric inner layer	<b>NBR</b>	2.2	HR	1.2
Fabric	Rayon	1	Polyester	1
Elastomeric outer layer	CR	1.3	<b>EPDM</b>	2.2
	$T_{\rm e}^{\circ}$ C		tan $\delta$	
	Virgin	After permeation test	Virgin	After permeation test
Hose no. 2				
NBR layer	$-44.0$	$-39.2+/-0.4$	0.37	$0.40+/-0.02$
CR layer	$-26.0$	$-21.4+/-0.3$	0.41	$0.35+/-0.01$
Hose no. 4				
IIR/EPDM layer	$-52.5$	$-51.0+/-0$	0.43	$0.46 \pm (-0.02)$
PA6/PA12 film	26.3	$34.5+/-1.1$	0.20	$0.17+/-0$

Table 9 Hose design and DMA results

the material recycling dominates in industrial practice today and probably in the foreseeable future. For this reason only this aspect will be treated here.

## 4.1. *SMC (sheet moulding compound) scrap for reinforcement of PA 66*

About 30% SMC was simply added without further pretreatment to polyamide. DMA measurements showed that only the onset temperature (the beginning of the glass transition range) of the storage modulus increased slightly to higher temperatures  $(8.2-12.4\degree C)$ . However, storage modulus below the glass transition range, 5.5 GPa for the SMC/PA compound compared to 4.4GPa, is on a distinctly higher level. At 1.5 GPa, as compared to l.OGPa, the average value above the glass transition range is also much higher. However, for temperatures up to about  $120^{\circ}$ C, the toughness, characterized by the mechanical loss factor tan  $\delta$ , is as much as 50% lower. This changes at higher service temperatures.

## 4.2. *Recycling of pure-grade thermoplastic production scrap*

This subject is actually state of the art in the plastics processing industry and is thus, solved if only a certain percentage of production scrap that has been processed several times is added to virgin material. For high performance and therefore very expensive thermoplastics like fluoroplastics, it is important to know whether it is possible to use 100% regrind without further compounding.

Two fluoroplastics were investigated, a tetrafluoroethylene/ethylene-copolymer (ETFE) from Du Pont (trade name Tefzel 280 MFR) and a polyvinylidene fluoride (PVDF) from Hills Troisdorf AG (trade name Dyflor 2000). Their price per kilo is about 80 DM, which means that the use of 100% regrind would be highly significant from the economical point of view. Because of their superb resistance to even highly aggressive liquids, which is a prerequisite for bottle dispenser heads for dosing such aggressive liquids in laboratories, fluoroplastics are used worldwide. In 1992 the firm Brand, Wertheim, sold 60,000 of its dispensettes<sup>®</sup> and the trend is still rising. In these dispensers, the core, the dispensette head as well as two valves are made of ETFE, the cannula-holding device for the output cannula is made of PVDF, and the suction pipe is made of tetrafluoroethylene/hexafluoropropylene-copolymer (FEP) [20].

ETFE and PVDT samples for tensile testing as well as finished parts like dispensette heads were produced on a Klöckner Ferromatik Desma FX50 microprocessorcontrolled injection-moulding machine. The material was thus processed once. Some of these test pieces were ground again and parts were injection-moulded once more from 100% of this regrind (2nd processing). The same procedure took place with some of these parts (3rd processing). After each processing stage, tensile testing samples were immersed for 30 days in standard laboratory chemicals such as toluene, tetrahy-



Fig. 10. Injection-moulded dispenser heads made of Tefzel 280 MFR: IR absorption quotient  $Q = A_1$  $(667 \text{ cm}^{-1})/A_2(509 \text{ cm}^{-1})$  versus number of processings.



Fig. 11. Injection-moulded dispenser heads made of Tefzel280 MFR: crystallization heat and maximum of the "crystallization rate" measured by DSC versus number of processings.

drofuran, acetone and concentrated hydrochloric acid. Then tensile strength, DSC and IR measurements were made. In the following, only ETFE will be discussed, because PVDF showed similar results. In addition, to tensile strength the crystallization temperature and the maximum rate of crystallization (maximum of the 1st derivative of the DSC signal in the crystallization zone), and the ratio of IR absorptions at 667 and 509 cm<sup>-1</sup>, which indicates the changes on the  $CF<sup>2</sup>$  groups of the molecule, proved to be significant values.

The crystallization was selected for the simple reason that it characterizes possible changes in the crystallization behaviour during injection moulding. If the DSC and IR investigations are examined, it can be seen that, within the measurement accuracy, no changes are displayed if the material has only been processed once (Figs. 10 and 11). Only after the material has been processed twice do slight changes of 7% occur in IR spectroscopy, which increases to 11% after the third processing (Fig. 10). Also only very small changes (max. 7%) could be found by DSC (Fig. 11). This is also verified by the tensile strength, even after immersion in chemicals (Fig. 12). No significant influence of the different chemicals was observed, so the conclusion is that 100% regrind from material processed twice can be used for a 3rd processing without problems.

## 4.3. *Identljication of plastics after* IO *years aging (service Ii&e)*

As far as quantity is concerned, this problem is surely of greater importance than the recycling of pure grade production scrap. One just has to think of the 14 million old cars in Western Europe or the 2.3 million in West Germany which were scrapped in 1991 [21]. The average service life of cars in Germany is 12 years. Therefore the surface of exterior automobile parts will be degradated by environmental influences to such a degree that it is no longer possible to identify plastics by IR spectroscopy without sample preparation. There are tests, however, in which if quick removal of the surface with a knife is sufficient to expose enough basic material, IR spectroscopy becomes possible [22]. Another method just as quick as IR spectroscopy but which does not make it necessary to remove the surface is photoacoustic spectroscopy (PAS) [23].



Fig. 12. Injection-moulded shoulder rods made of Tefzel 280 MFR: tensile strength after 30 days.



Fig. 13. Photoacoustic spectrum of polybutyleneterephthalate after ten years aging under four different conditions.

By selection of the appropriate modulation frequencies, it is possible to receive acoustic signals (heat waves) from different depths of the test piece from IR radiation absorbed by the sample. The quality of the spectra received is even better if the surface is structured, as in the case of foam, for example. A PA 66 (trade name, Ultramid A3K from BASF), a polycarbonate PC (trade name, Makrolon 2800 from Bayer AG) and a PBT (trade name, Ultradur B 4500 from BASF; Fig. 13 shows the PAS spectra) were measured after they had been stored for 10 years at 23°C and 50% relative humidity, at 55°C and 20% relative humidity, at 40°C and 92% relative humidity, or outdoor weathering at an angle of  $45^{\circ}$  in southerly direction in the area of Munich.

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